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Citation: Applied Physics Letters **99**, 073103 (2011); doi: 10.1063/1.3624590 View online: http://dx.doi.org/10.1063/1.3624590 View Table of Contents: http://scitation.aip.org/content/aip/journal/apl/99/7?ver=pdfcov Published by the AIP Publishing

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Enhanced gas-flow-induced voltage in graphene

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(Received 4 July 2011; accepted 22 July 2011; published online 15 August 2011)

We find experimentally that gas-flow-induced voltage in monolayer graphene is more than twenty times of that in bulk graphite. Examination over samples with sheet resistances ranging from 307 to 1600 Ω /sq shows that the induced voltage increases with the electric resistance and can be further improved by controlling the quality and doping level of graphene. The induced voltage is nearly independent of the substrate materials and can be well explained by the interplay of Bernoulli's principle and the carrier density dependent Seebeck coefficient. The results demonstrate that graphene has great potential for flow sensors and energy conversion devices. © 2011 American Institute of Physics. [doi:10.1063/1.3624590]

Use of wind power and creation of self-powered devices are most attractive issues in both academic and industry fields.¹⁻³ Gas flow passing the surface of a conductive material can generate a voltage due to the interplay of Bernoulli's principle and the Seebeck effect.⁴ Graphene, as the thinnest material with extraordinary electrical properties such as flexibility and strong mechanical strength,^{5–9} can be split off from bulk graphite by mechanical or chemical cleavage methods,^{10,11} directly grown on large scale on surfaces of many materials.^{12–15} It can be transferred to various substrates, transforming the surfaces from insulator into conductor^{14–17} and protecting the surfaces from oxidation and corrosion.¹⁸ If all these promising properties can be merged with gas-flowinduced voltage, graphene will have great potential for velocity measurement, wind energy conversion, and creation of selfpowered devices. However, the bulk graphite shows poor ability to yield voltage from gas flow.⁴ Here, we find that the single layer graphene can enhance the ability in produce voltage from gas flow by twenty-fold over the multilayered graphite, nearly independent of the substrates. This intriguing ability can be further improved by controlling the quality and carrier concentration, thus opens potential for flow sensors, self-powered devices, and energy conversion.

Single layer graphene samples used in this work were synthesized by recently developed chemical vapor deposition (CVD) method.^{13–15} After CVD growth on the copper foil, the graphene was firstly transferred on to a 1 mm thick quartz substrate.¹⁷ A photo of a graphene sheet of $5 \times 25 \text{ mm}^2$ in size transferred onto the quartz substrate is shown in the inset of Fig. 1(a). The transferred graphene samples were then characterized by a Renishaw Raman spectrometer with solid-state laser ($\lambda = 514.5$ nm), as shown by Fig. 1(a). The well symmetric 2D peak shape with full width at half-maximum (FWHM) $\sim 31 \text{ cm}^{-1}$, small ratio of G to 2D peak intensity (*I*(G)/*I*(2D)) < 0.5), and negligible disorder induced D peak clearly show that the sample is monolayer graphene with low density of defects.^{19,20} The blueshifted position of G peak (\sim 1590 cm⁻¹) for our sample compared to that of exfoliated graphene $(\sim 1586 \text{ cm}^{-1})$ indicates relatively higher carrier concentration $(>3 \times 10^{12}/\text{cm}^2)$.²¹ The gas-flow-induced voltage (V_{GF}) of the sample was measured using devices consisting of uncovered graphene sheet, copper electrodes, and directed flow pipe as schematically illustrated in Fig. 1(b). The electrical contacts at both ends were made with copper foil adhered by silver emulsion. The exposed part of the graphene sample was adjusted to 10 mm in length along the flow direction and 5 mm in width. The outlet of gas flow tube is kept 5 mm away at an angle of $\alpha = \pi/4$ with respect to the graphene surface to obtain a maximum output voltage signal. The sheet resistance of the sample is measured to be 943 Ω /sq. The velocity of the gas flow (ν) was measured using a rotameter and a digital pressure indicator. The induced voltage was measured using a KEITHLEY 2010 multimeter.

Figure 1(c) shows the typical voltage response from the above setup system as a function of time when the gas flow is switched on and off. The response of a graphite sample (highly oriented pyrolytic graphite, HOPG) is also displayed for comparison. When an Ar flow with velocity of 68 m/s is turned on, the induced voltage signal increases from zero to $-58 \ \mu\text{V}$ and 2.5 $\ \mu\text{V}$ for graphene and graphite, respectively. Apparently, the signs of V_{GF} for graphene and graphite are opposite as they are in *p*- and *n*-type doping states respectively, and the magnitude of V_{GF} for graphene is over twenty-fold larger than that for graphite of comparable size. As the gas flow is turned off, the voltage decreases and eventually returns to zero. The alternating on-off gas flows produce a series square-wave-like voltage signals in the samples and the voltage response to the gas flow is within a few seconds.

To examine the influence of sample quality of graphene for the flow induced voltage, we performed the measurement with 12 CVD graphene samples transferred onto quartz substrate, with sheet resistance values ranging from 425 Ω/sq to 1.6 k Ω/sq . Figure 2(a) shows the variation of V_{GF} for the graphene devices against σ , the reciprocal of sheet resistance. It can be seen that the V_{GF} presents slightly increasing tendency with the value of sheet resistance. According to the theory developed in carbon nanotube and other materials,⁴ the V_{GF} can be deduced as $V = kS\Delta T$, where S is the Seebeck coefficients of the sample material with respect to the electrode material, ΔT is the gas flow induced temperature difference along

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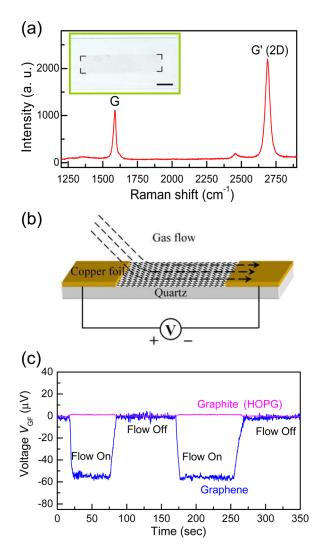


FIG. 1. (Color online) (a) Typical Raman spectra of single layer graphene on quartz substrate, the inset shows a photo of graphene sample, bar = 5 mm. (b) The schematic illustration of the gas flow experiment setup. (c) Typical voltage signal response of graphene and graphite for switching on/ off the argon gas flow at 68 m/s.

the sample, and k is a fitting factor. In the case of graphene devices, the negative sign of $V_{\rm GF}$ suggests a *p*-type doping state, in consistent with literature reports that graphene can be easily *p*-doped via absorption of water molecular or oxidative ions onto its surface in ambient air.^{22–24} The divergent values of $V_{\rm GF}$ recorded at the same gas-flow condition suggest the different Seebeck coefficients for various graphene samples.

The Seebeck coefficients of graphene is strongly dependent on the carrier density *n*. Theoretical calculations based on experimental results^{25–27} and the effective-medium theory²⁸ show that the absolute value of *S* for graphene peaks at $n \sim 1 \times 10^{12}$ /cm² and decreases with further increase in *n*. In high density region, the Seebeck coefficient of graphene behaves as $1/\sqrt{n}$, in agreement with the semiclassical Mott formula. As the transferred CVD graphene typically has carrier density larger than 1×10^{12} /cm² due to the doping of charged impurities in the transfer process,¹⁵ which is also confirmed by our Raman characterization, and the lower sheet resistance of graphene film implies higher carrier density. The decreasing tendency of $V_{\rm GF}$ against increasing σ in Fig. 2(a) can be well explained by the density dependence of Seebeck coefficient in graphene.

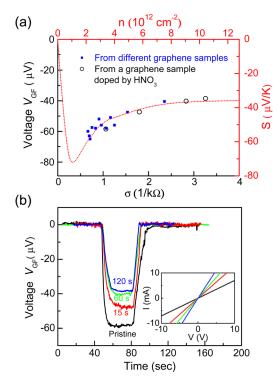


FIG. 2. (Color online) (a) Voltage response for graphene samples with different resistances (blue squares) and a graphene sample with different HNO₃ treating durations (black circles). Doted line shows the predicted Seebeck coefficient *S* as a function of carrier concentration at 300 K following the effective-medium theory.²⁸ (b) Voltage response of pristine and the HNO₃ treated graphene samples in argon gas flow at 68 m/s. Inset shows *I-V* curves of the samples.

The carrier density of graphene can be effectively controlled by doping level in graphene. To modulate the concentration of *p*-type carriers (holes), the graphene sample was treated in a nitric acid (HNO₃) vapor environment¹⁵ for 15 s, 60 s, and 180 s; the corresponding I-V curves for the pristine and treated samples are shown in the inset of Fig. 2(b). The sheet resistance reduced from 943 Ω /sq for pristine graphene to 307 Ω /sq for the sample with 180 s treatment. Following each HNO₃ treating process, V_{GF} measurements were conducted under an argon gas flow at velocity of 68 m/s. The square-wave-like voltage responses are shown in the inset of Fig. 2(b). The deeper in *p*-doping of the graphene sample leads to lower resistance and a weaker response than the pristine sample. The induced voltage decreases from 58.5 μ V to 38.5 μ V gradually. The V_{GF} against σ data from our experiments, the calculated S against n curve by the effective-medium theory (near charge-neutrality point), and Mott formula (high carrier density region)²⁷ are plotted in Fig. 2(a), where the experimental data shows nearly the same tendency as the theoretical curve. Note that although V_{GF} is demonstrated to be reduced by HNO₃ vapor treatment here, it is possible that $V_{\rm GF}$ can be further enlarged by continuous reducing the carrier density along the theoretical curve.

The gas-flow-induced voltage V_{GF} of various materials is strongly flow-velocity-dependent. The dependence of voltage signal on the square of Mach number M for graphene and graphite is shown in Fig. 3(a), where M = v/c, v is gas flow velocity, and c is the sound velocity in medium (323 m/s for argon at 300 K). It can be seen that the value of V_{GF} increases linearly with M^2 , with slop of -1325 for graphene

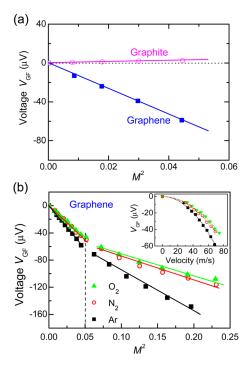


FIG. 3. (Color online) (a) Ar gas flow induced voltage V_{GF} versus M^2 for graphene and graphite. (b) Variation of V_{GF} against M^2 for argon, nitrogen, and oxygen gases. Solid lines are linear fitted to the data. Inset of (b) shows the dependence of V_{GF} on velocity at low velocity region.

and 56 for graphite. Due to Seebeck effect, the V_{GF} can be produced by the change in temperature along the gas flow direction on sample surface. According to Bernoulli's principle, an increase in the speed of the gas flow occurs simultaneously with a decrease in pressure, which in turn causes a temperature increment. For adiabatic flow of a gas, the temperature difference can be approximatively deduced as $\Delta T \propto \frac{1}{2} T_0 \gamma M^2$ (Ref. 4) at low velocity region ($M^2 < 0.5$) and $\Delta T \propto \frac{1}{2}T_0(\gamma - 1)M^2$ at high velocity region ($M^2 \ge 0.5$), where T_0 is the environment temperature and γ is the heat capacity of gas. Thus, in low velocity region in Fig. 3(a), $V_{\rm GF} \propto S\Delta T \propto \gamma S \cdot M^2$. The dramatic increase in $V_{\rm GF}$ for graphene than its bulk form graphite can be attributed to two reasons: (i) The Seebeck coefficients of graphene, typically $30 \sim 60 \ \mu\text{V/K}$ in recent reports^{22–25} at room temperature, is much larger than that of graphite ($-8 \mu V/K$). (ii) As atomic thin material, graphene can be most sensitive to the surface temperature change. The flow-velocity dependence of V_{GF} for graphene in argon, nitrogen, and oxygen gases exhibits similar linear tendency (Fig. 3(b)), except the slightly different slopes which are mainly attributed to the difference in γ value [γ (nitrogen): γ (oxygen): γ (argon) = 1:1:1.2]. The lower slops at high velocity region ($M^2 > 0.05$) are attributed to the density changes of the gas.

As a flexible, stable, and easy-processable material, graphene can be transferred to varied substrates including polymers and crystal wafers. To find the influence of substrates, we measured the gas flow induced V_{GF} of graphene samples of the same size on different substrates under argon gas flow of 68 m/s. The average output signal is measured to be 53 μ V for polytetrafluoroethylene (PTFE), 50 μ V for polymethylmethacrylate (PMMA), 53 μ V for polyethylene terephthalate (PET), 55 μ V for quartz, 51 μ V for mica, and

54 μ V for 300 nm SiO₂/Si substrates. As the measurement error is about ±4 μ V, the measured range for different substrates suggests that the V_{GF} response of graphene is nearly independent of the insulating substrates.

In conclusion, we have shown that the gas flow induced voltage in graphene can be more than twenty times of that in bulk graphite, owing to its atomic-thin structure with high Seebeck coefficient. The induced voltage shows increasing tendency with the sheet resistance, which can be explained by the carrier density dependent Seebeck coefficient. The high ability of graphene to produce voltage under gas flow is proven to be robust to substrate materials and can be further improved by controlling the quality and doping level. The results exhibit that graphene can be a promising candidate for gas flow sensors and high efficient energy conversion devices.

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